Dr. Nandini Devi NCL





Anu Aga and family







States of Matter & Crystal Growing





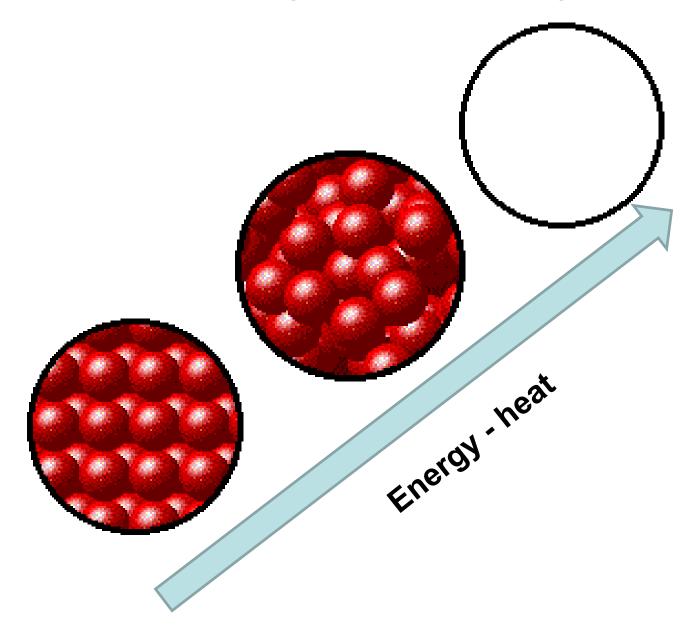








How are the molecules arranged in solids, liquids and gases?



What are the components of air? Oxygen, Nitrogen, Argon, Carbon dioxide etc.

Nitrogen – N₂

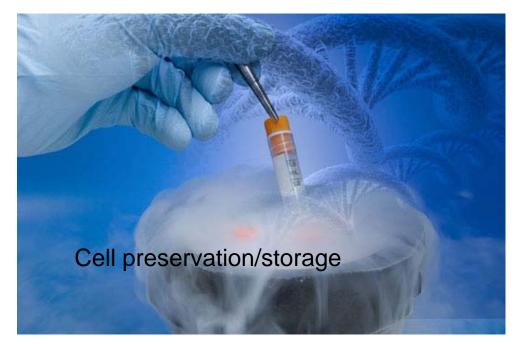
How is liquid N₂ made?

If the temperature is brought down to -196 °C!

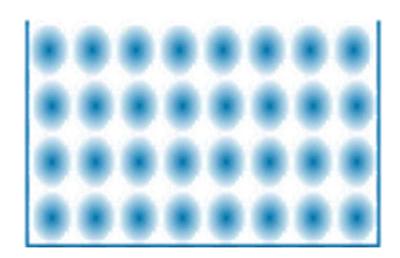


Such processes are called **cryogenics**

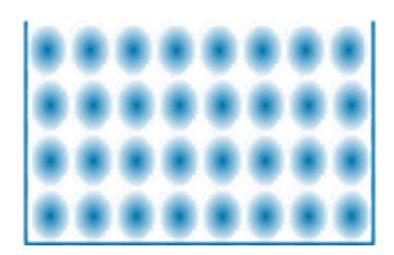


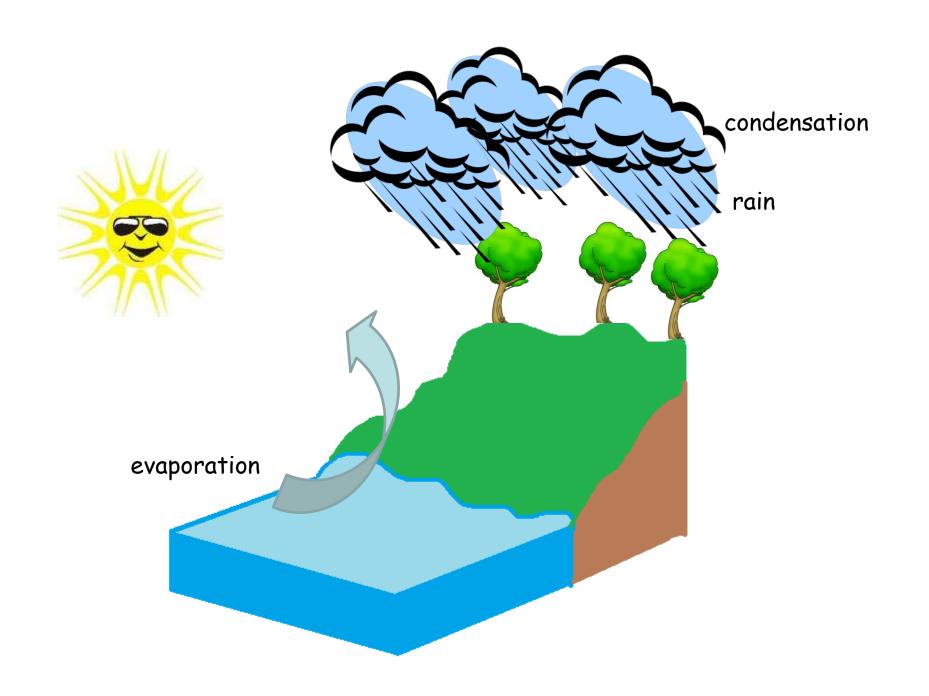


Vapourisation: when the liquid becomes gas at its boiling point



Evaporation: when the liquid becomes gas below its boiling point





Carbon dioxide – CO₂

How is solid CO₂ made? If the temperature is brought down to -78.5 °C!

Sublimation: when solid becomes gas directly

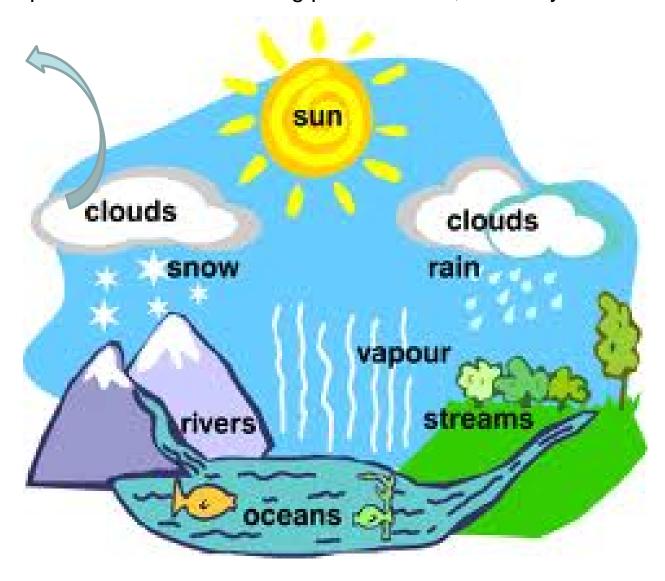


Frozen food transport

Organ transport



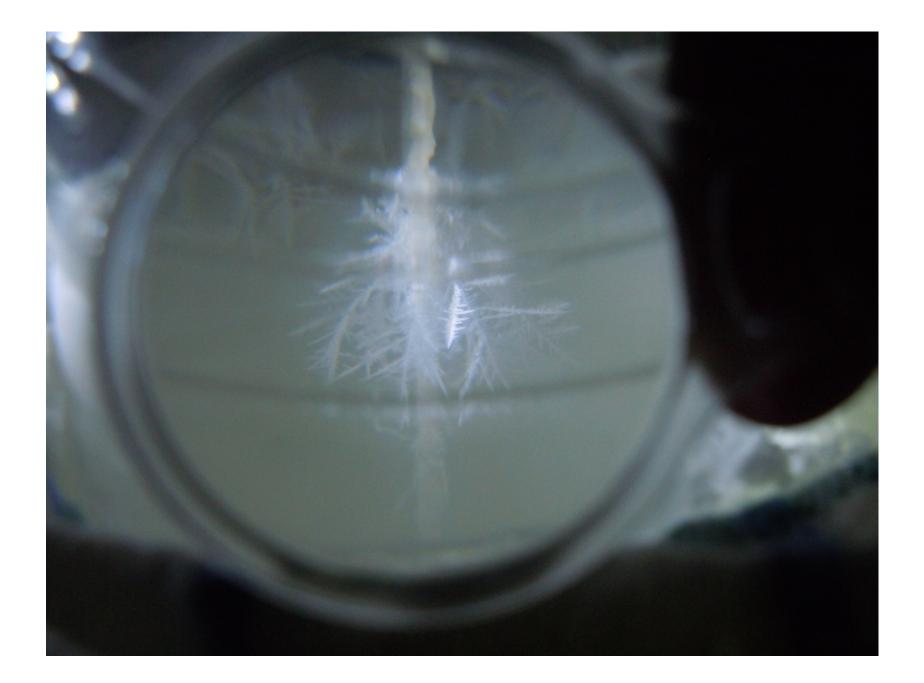
When the temperature is below freezing point of water, small crystals of ice form





Let us make snow!!!

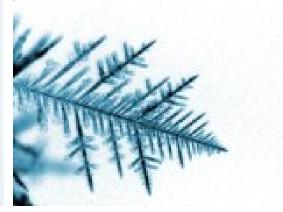


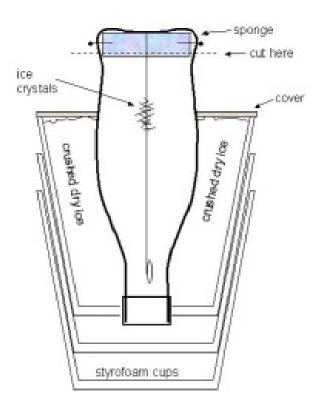


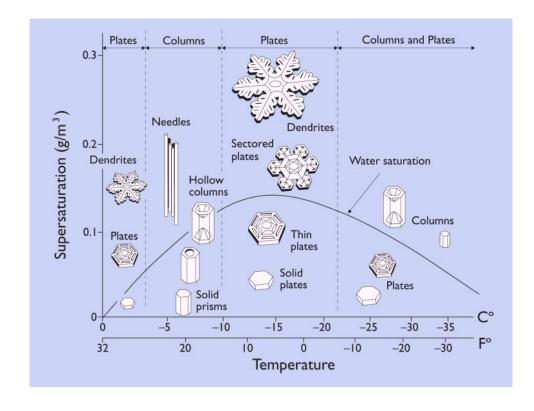
Dendritic snow crystals formed inside the bottle



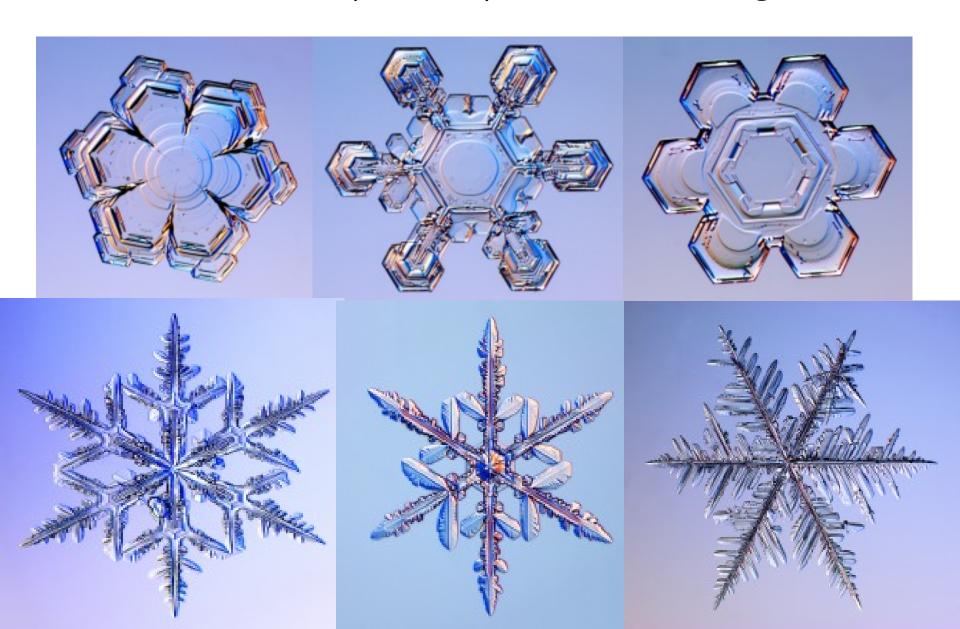




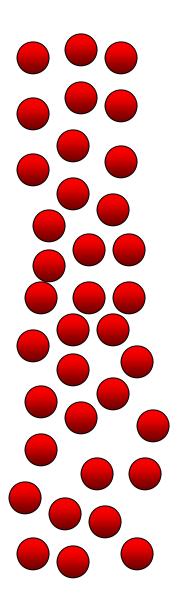




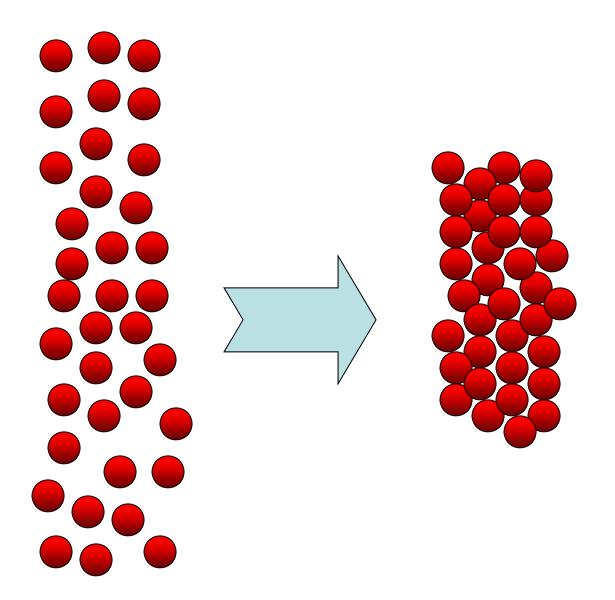
Natural snow crystal shapes - 6 sided hexagons



Water vapour forming snow – very ordered - crystallline



Solid but but no order - amorphous



Solids - crystalline and amorphous





quartz sand

Both are chemically more or less same based on silica SiO₂

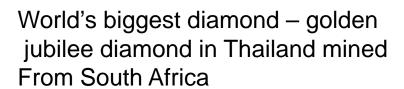
Solids - crystalline and amorphous



charcoal diamond

Both are chemically same based on carbon C







Universe's biggest diamond – star Lucy In constellation Centaurus, 50 light years from Earth!!!

Crystals!!!!! Nature has made beautiful crystals



beryl

How did these beautiful crystals form in nature?

Geode – a natural pressure cooker!





Earth cooled from molten state

Took millions of years!!





Very very big crystals of gypsum (magnesium sulphate) in a cave in Mexico

Let's make some cool crystals - much faster than nature!!

Crystallisation from supersaturated solutions

Epsom salt, sodium carbonate, alum



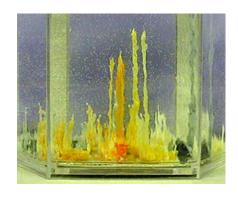
Alum
Aluminum sulphate

Crystallisation by reactions

Silver

$$Cu + AgNO3 = Cu(NO3)2 + Ag$$

Rock garden in a bowl!



Na2SiO3 + M(NO3)2 = MSiO3 + 2NaNO3

Crystallisation from melt cooling Thiosulphate

